

Measuring the Viscosity of Gases with a Greenspan Viscometer

J.J. Hurly^{C, S}

Chemical Science and Technology Laboratory, Process Measurements Division, National Institute of Standards and Technology, Gaithersburg, MD, U.S.A.

K.A. Gillis

Chemical Science and Technology Laboratory, National Institute of Standards and Technology, Gaithersburg, MD, U.S.A.

The Greenspan viscometer is an acoustic resonator composed of two gas-filled chambers coupled by a duct of radius r_d . In the lowest order, the viscosity of the gas is $\eta = \pi f \rho (r_d/Q)^2$, where f and Q are the resonance frequency and quality factor of the isolated Greenspan mode, and ρ is the gas density. We installed a Greenspan viscometer in an automated facility capable of spanning the temperature range 200 K to 400 K and pressures up to 3.3 MPa. We measured the frequency and Q of the Greenspan mode when the resonator was filled with seven gases: Ar, He, CH₄, C₃H₈, N₂, SF₆, CF₄, and C₂F₆. The data were interpreted using a complete, parameter-free model [1]. In the limit of low density, the resulting values of the viscosities of Ar, He, N₂, C₃H₈ and SF₆ agreed within ± 0.3 % with the values obtained by Vogel *et al.* [3-4] using oscillating disk techniques. The present results and Vogel's results are consistent with the viscosity ratios determined by Berg using capillary flow [5].

- [1] K. A. Gillis, J. B. Mehl, M. R. Moldover, Submitted to *J. Acou. Soc. Amer.*
- [2] E. Vogel, *Ber. Bunsenges. Phys. Chem.* 88, 997 (1984).
- [3] E. Vogel, C. Kuechenmeister, E. Bich, and A. Laesecke, *J. Phys. Chem. Ref. Data* 27, 947 (1988).
- [4] T. Strehlow and E. Vogel, *Physica A* 161, 101 (1989).
- [5] R. F. Berg, Private Communication.